

Date: ORGANIZATION AND MAINTENANCE OF ORGANISMS

(Basic Understanding of Topic)

Nutrients / Biological molecules:

⇒ Major nutrients:

- carbohydrates
- proteins
- fats / lipids
- nucleic acids
- H<sub>2</sub>O

⇒ Minor nutrients:

- minerals
- vitamins
- inorganic chemicals

★ CARBOHYDRATES:

→ organic chemicals

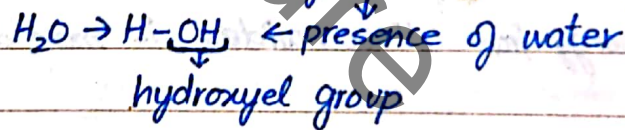
→ ①. Carbon ②. Hydrogen ③. Oxygen

→ 1 : 2 : 1

→ general formulae: (CH<sub>2</sub>O)<sub>n</sub>

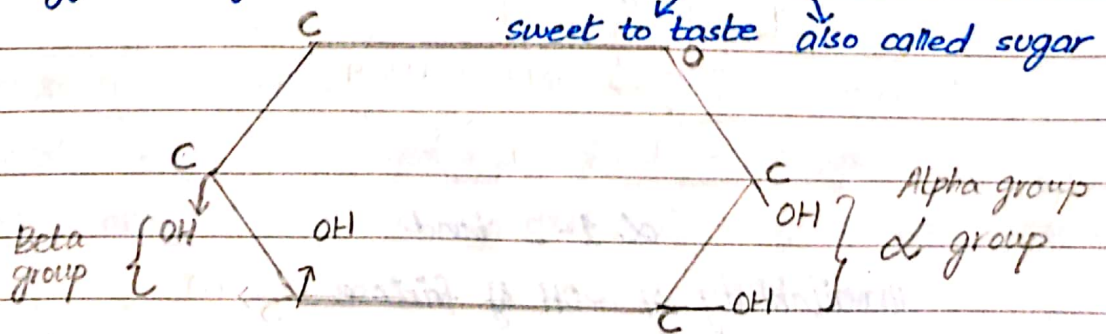
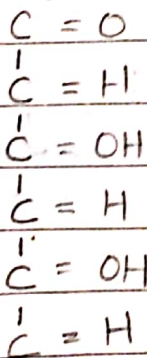
→ Example: glucose C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> the no. 6

→ Carbohydrate's name comes from the name hydrated carbon



→ every carbon must have water

→ In greek terminology carbohydrates are called saccharides.



→ soluble in water. solubility depends on simplicity.

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- e.g:
- glucose - readily soluble
  - starch - slightly soluble
  - cellulose - insoluble in water

→ 3 basic types:

①. MONOSACCHARIDES:

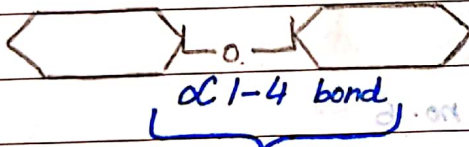
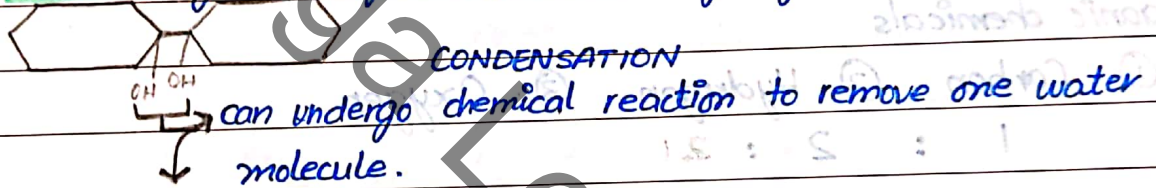
one unit

- Example:
- glucose  $C_6H_{12}O_6$   hexagon
  - fructose  $C_6H_{12}O_6$  (reducing sugar)  pentagon

②. DISACCHARIDES

Two units contains double energy as compared to monosaccharides.

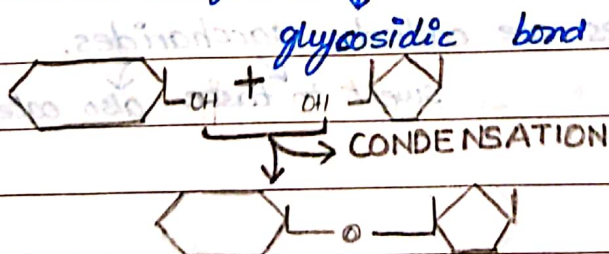
- Example: ● maltose (glucose - glucose) ⇒ reducing sugar



Glycosidic Bond

→ To gain energy from disaccharides is more difficult than to gain from monosaccharides

- sucrose (glucose - fructose)



$\alpha$  1-2 bond

unavailability of -OH of fructose

Sucrose - Non-reducing Sugar

Glucose  $\rightarrow$  liver  $\rightarrow$  stored in liver cells  
 (required) Blood  $\rightarrow$  to glucagon

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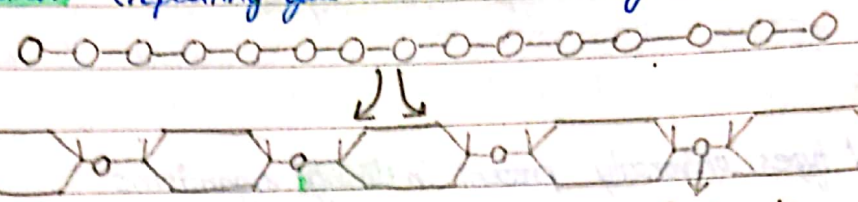
happens so that they won't disturb water potential of liver cells or continuous endosmosis resulting in cell bursting because it is

③ POLYSACCHARIDES

multiple or more than two

insoluble, more energy levels because

Example:  $\therefore$  starch (repeating glucose units making a chain) (plant chemical) of the complex nature

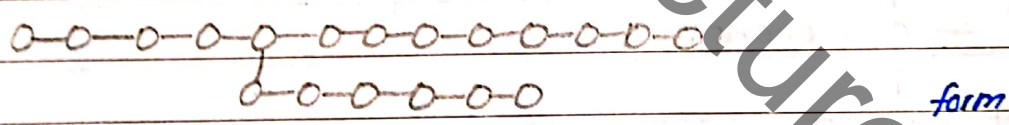


no matter how many units bond will be of 1-4

- $\Rightarrow$  used in plants as storage carbohydrate.
- $\Rightarrow$  more bonds in polysaccharides = more storage space
- $\Rightarrow$  preferable form of starch as would not finish its entity by dissolving in cytoplasm disturbing its water potential.

$\therefore$  glycogen (repeating glucose units) (animal starch)

- $\Rightarrow$   $\alpha$  1-4 bond
- $\Rightarrow$  branching chains of glucose every 10-12 units



$\Rightarrow$  because of the chain type it becomes a storage<sup>form</sup> for carbohydrates in animals.

$\Rightarrow$  insoluble in cytoplasm - for breaking  $\beta$  1-4 bond<sup>enzyme</sup>

herbivores can digest due to presence of cellulase

$\therefore$  cellulose (repeating glucose units) (plant chemical)

(humans can't break  $\beta$  1-4 bond) (remains undigested in humans)

$\Rightarrow$   $\beta$  1-4 bond making it more complex and rigid and cell wall harder.



$\Rightarrow$  can contain high levels of water due to the rigidity of cell wall.

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# PROTEINS

- ⇒ long chains of molecules amino acids
- ⇒ compose carbon, hydrogen, oxygen, nitrogen

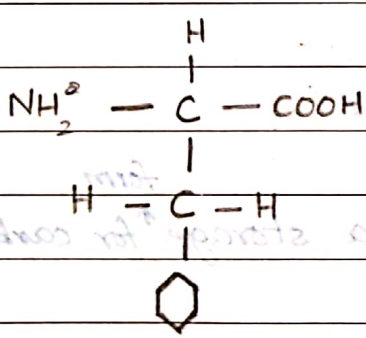
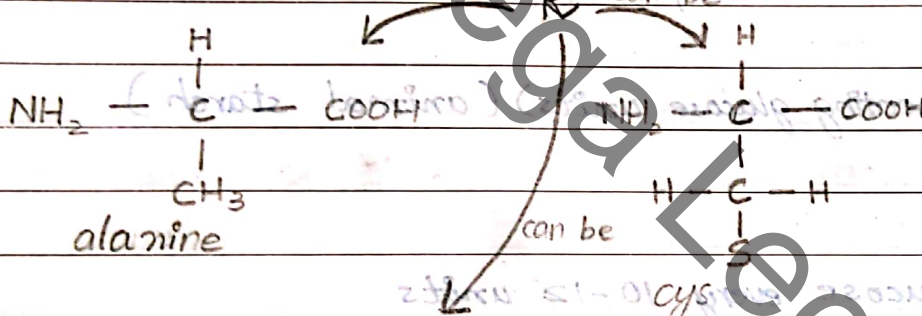
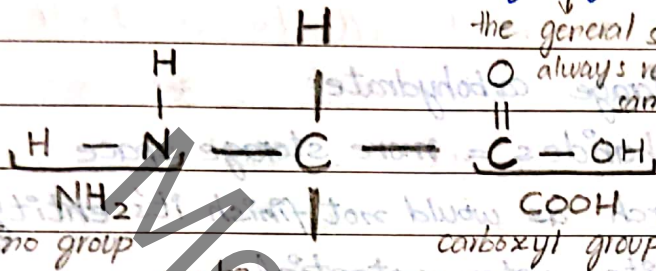
## AMINO ACIDS

though there is specie-specific charges : - sequence of amino acid

⇒ 22 different types universally present in living organisms

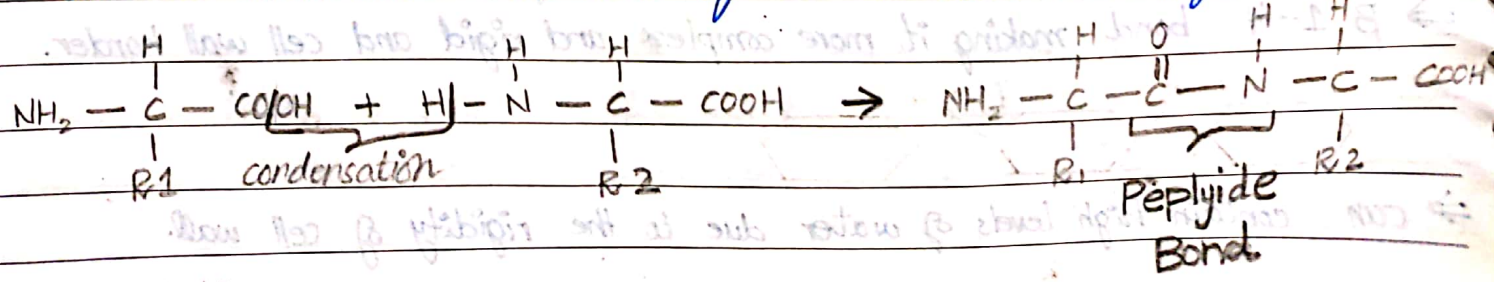
- no. of amino acids

↑  
the general structure always remains the same



R : causes the variations : 22 types

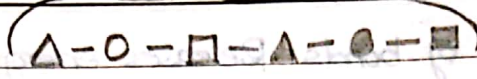
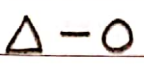
⇒ whenever amino acids come side by side to each other they form bonds.



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larger the chain the more the no. the more energy is stored the number can vary.

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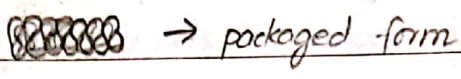


Dipeptide chain

Polypeptide chain

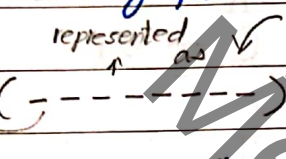
no. of amino acid bonds are present.

⇒ polypeptide chain is packaged / compacted / folded for storage and to make a mature protein.



⇒ When folded they form hydrogen bonds. Positive charge on end and negative on another. can break very easily. / keeps the chain stabilized /

high pH, temp., concentra., can loose hydrogen bonds. (enzyme denaturation)



loose attractive bonds

⇒ the folding causes the polypeptide to have a shape which can be different from others.

- enzyme - active site
- hormone - receptor site / binding site
- HB - oxygen binding site

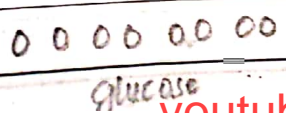
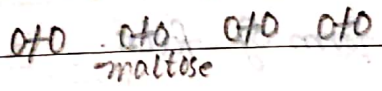
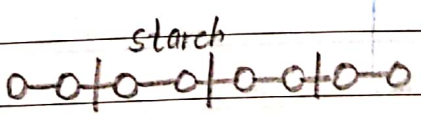
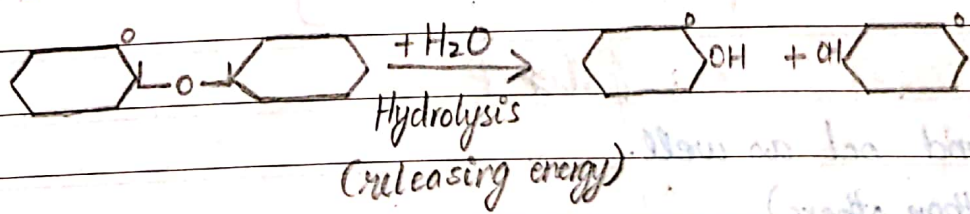
folding is

⇒ not only important for stabilizing but also binding shapes. and for maturity of the protein

⇒ Proteins are hence, forming a shape which whose function is imp. to the body.

II Reversal / Breakage of Bonds

Glycogen Bond:



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**HYDROLYSIS**: Breakage of bonds. Releases energy

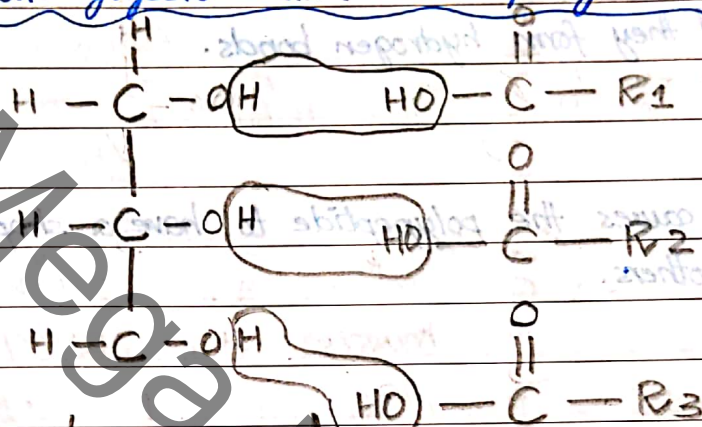
**CONDENSATION**: Bonds are made. Energy is stored

**FATS**

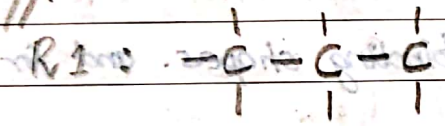
→ storage chemibiological molecules, energy reserves for future use

→ made of carbon, hydrogen and oxygen

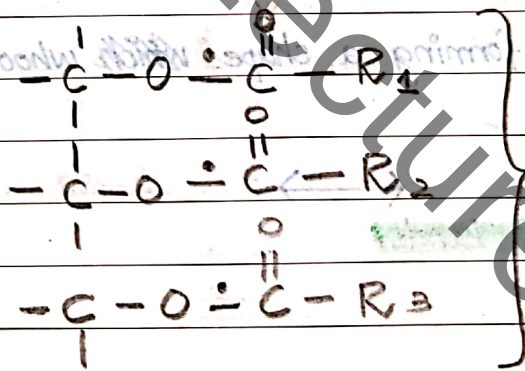
→ are a combination of one glycerol and three fatty acids



E.g.//



- 3 H<sub>2</sub>O  
Condensation



BASIC  
STRUCTURE  
OF  
FATS

• : Ester Bond

- ⇒ called lipids and ocl as well.
- complex (more than others)
- difficult to break down
- conserves more energy
- human body favours to store energy in the form of fats.
- lower abdomen has more fat deposition

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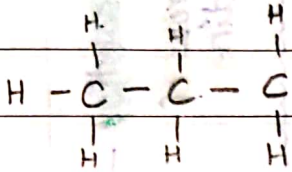
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∴ SATURATED FATS

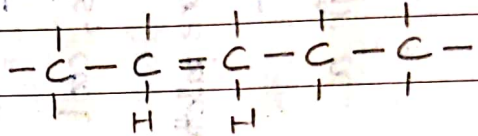
→ single bonds in carbon R chains

∴ UNSATURATED FATS

→ double or triple bonds in carbon R chains



SATURATED



UNSATURATED

oil

HYDROGENATION

Add hydrogens

not advised  
as moves as  
semi-solid. makes  
a disturbance to

advised to use oil as  
moves as a liquid and  
is more convenient for  
blood vessels.

the walls of vessels increasing blood pressure.  
also would disturb the contents of blood plasma. } making clots which can block vessels

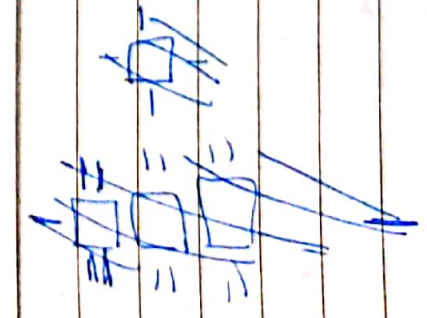
→ larger in size than glucose & amino acids

TEST	OBSERVATION	RESULTS
<p>①. <b>Iodine Test / Starch solution</b>                      2cm<sup>3</sup> of starch sol. + 3-4 drops of (yellowish brown) iodine</p>	<p>→ Blue Black Colour                      → Brown Colour</p>	<p>Positive starch test                      Negative starch test</p>
<p>②. <b>Reducing Sugar Test</b>                      2cm<sup>3</sup> of glucose sol. + 2cm<sup>3</sup> Benedict's (doesn't act in acidic) reagent (medium only in neutral)                      + heat in water bath for 3-5 minutes (for indirect heat)</p>	<p>→ Green → Yellow → Brown → Orange                      Beet Red ← Brown                      (mild appearance if concentration is less)                      → Blue colour remains same</p>	<p>Positive sugar test                      Negative Test</p>
<p>③. <b>Non-Reducing Sugar Test</b>                      2cm<sup>3</sup> of sucrose sol. + dilute HCL (can perform hydrolysis in presence of enzymes) + 2cm<sup>3</sup>                      + heat for 3-4 minutes, cool at room temperature + sodium hydroxide (solid) one palette + Benedict's reagent + heat again for 3-4 min. in a water bath</p>	<p>→ Blue → Green → Yellow → Brown                      Beet Red ← Brown ← Orange</p>	<p>Positive Non-Reducing Sugar</p>

e.g) glucose, fructose, maltose, lactose

e.g) sucrose


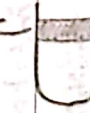
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Water is an inorganic solvent

T E S T	O B S E R V A T I O N	C O N C L U S I O N
<p>④. Biuret Test / Protein Test</p> <p>2cm<sup>3</sup> of Egg Albumin + 2cm<sup>3</sup> of sodium hydroxide + 3-4 drops of CuSO<sub>4</sub> in a standing table</p>	<p>→ Violet coloured ring</p> <p>→ Blue</p> 	<p>Positive Protein Test</p> <p>Negative Protein Test</p>
<p>⑤. Emulsion Test / Fats Test</p> <p>1cm<sup>3</sup> of sample oil + 1cm<sup>3</sup> of alcohol ethenol / ether (organic chemical) + shakeeee + distilled water 1cm<sup>3</sup></p>	<p>→ White emulsion / suspension appears at the top</p> <p>→ Clear solution</p> 	<p>Positive fats Test</p> <p>Negative Fats Test</p>

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- PROVIDED THINGS:
- 5 Test Tubes
  - One stand
  - All the Chemicals

Date:

## MINOR NUTRIENTS

### MINERALS

- not produced by the living organs
- chemicals which are required, absorbed by plants their roots
- intake of plants, they enter the organisms body.

E.g//

- **Nitrogen** (present in form of salts, nitrates, nitrites)  $(NO_3^-)$   $(NO_2^-)$   
↓  
lack causes (in blood absorbed by) become proteins → growth  
stunted growth

- **Magnesium** (central chemical of chlorophyll)  
↓  
lack results in chlorosis presence affects photosynthesis  
yellow leaves

- **Iron** (central chemical of HB) e.g/ liver, red meat, beans & nuts, dried fruit  
↓  
deficiency results in decreased increasing  $O_2$  carrying capacity of blood  
oxygen in blood } anemia  
anemia

• Deficiencies can be recovered by increasing the intake.

- **Phosphates** (used to make nucleic acids)  
 $(PO_4^-)$  genetic mol. (DNA/RNA)

- **Calcium** (Ca deposits in bones and teeth and makes it strong) e.g/ cheese, eggs, milk, leafy green vegetables  
↓  
specifically required by animals on different levels also imp. for muscle contraction  
also imp. for nerve impulse

⇒ All are required in minor quantities but on regular level. Their lack can cause ... and can result in diseases.

Date: \_\_\_\_\_

## VITAMINS

- organic chemicals
- not produced by animals, produced by plants
- required by animals though

E.g //

- **Vitamin A** (source is carrot and citrus fruits)  
↓ produces retinol → imp. for improvement of vision

lack (extreme) results in night blindness,  
weak eyesight

- **Vitamin C** (source is egg, milk and meat and citrus fruits)  
↓ lack of VC causes bleeding gums, pale skin  
↓ scurvy gums  
→ also called Ascorbic acid  
imp. for our normal gums and skin

leafy green vegetables

- **Vitamin D** (source is liver, fish liver oil (cod-liver oil), eggs and fish and sunlight)  
↓ also known as calciferol

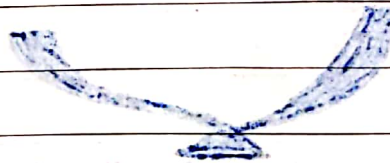
lack in infants will cause (reversible)  
weaker, soft and bending bones.

imp. for movement  $Ca, PO_4$ . it makes  
bones stronger. No Vitamin D and presence  
of  $Ca, PO_4$  is no use

Rickets

- direct exposure of skin to the sunlight

⇒ animal bodies can not store vitamins. fresh supply of vitamins.



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→ Herbivores at the level of appendix have certain bacteria that digest cellulose. It produces the enzyme cellulase

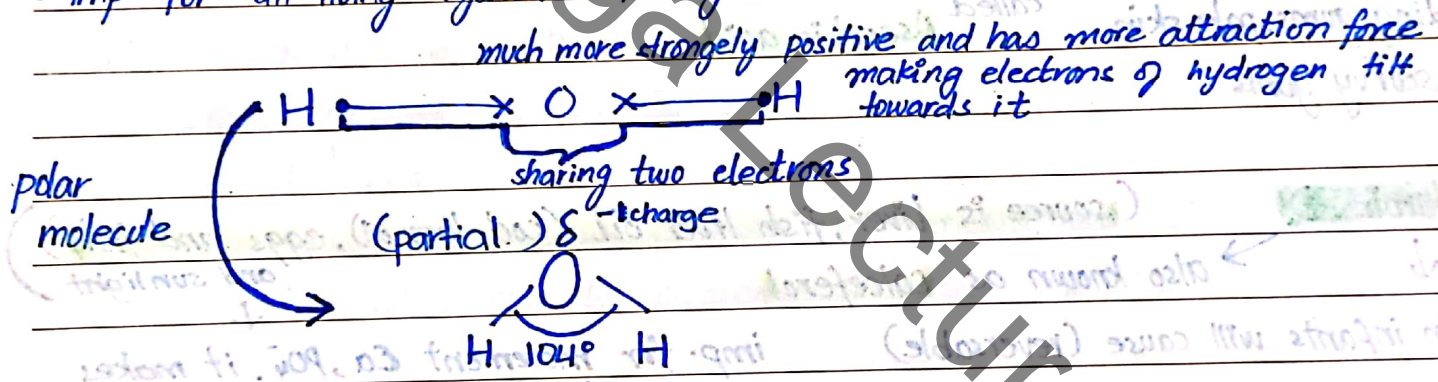
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## FIBER

- also termed as Roughage
- undigested waste material (plant food)
- presence of water in plants make faeces soft which lets it pass through the last part of digestive system
- lack of plant intake — constipation (damages wall of last part)
- undigested cellulose also blood
- it can absorb water from digestive track which makes it soft
- minor quantity e.g. fruit, vegetables & cellulose

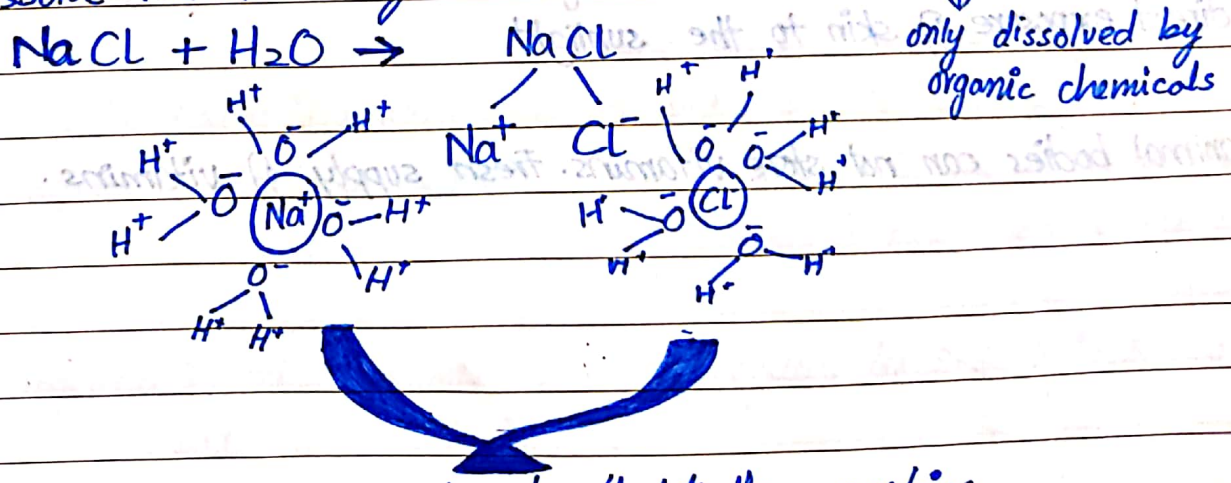
## WATER

→ imp. for all living organisms + single cell



→ polar characteristic makes it an universal solvent.

→ can dissolve maximum of the solute (except fats and oils)



water doesn't let them combine

Date:

- required for all metabolic reactions in cells
- required to transport materials in and out of cells
- required to transport materials within the body of a multi-cellular organism. E.g.:
  - minerals from roots to leaves
  - nutrients from digestive system to all cells
- required to maintain body temperature. E.g.:
  - sweating in animals
  - transpiration in plants
- acts as a reactant for photosynthesis
- maintains shape of plant cells and internode of the plant body.  
known as Hydroskeletal
- is involved in movement of parts of the plant. E.g.:
  - opening + closing of flowers
  - leaves towards the

sources of water:

↳ food & drinks

↳ metabolic processes e.g./aerobic respiration

• sunlight

★ with reference to energy do not use production of energy. only release

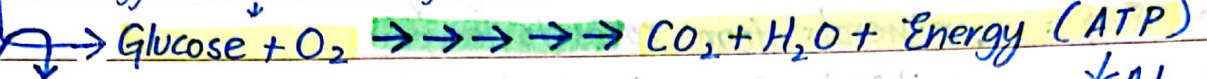
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### FOOD AND ENERGY

Living cells → Mitochondria

(for the formation of ATP (the energy released here is enough)) ↓

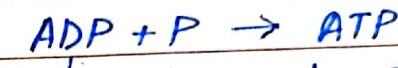
• energy conserving chemical



↓ Adamin Triphosphate (required at the level of active transport)

all are food materials should be brought to level of glucose

→ Carbo., Proteins, Fats

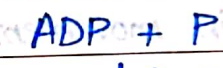


↓ Adamine DiPhosphate + Phosphate

carry less energy than proteins & carries energy less than fats.

• the bond is broken to consume energy

fats because of their simple nature

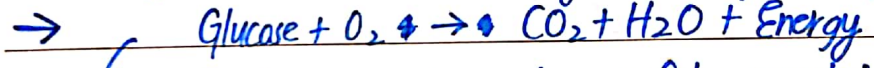


Release ↓ ↑ Conservation ATP

→ Fats are energy storing molecules. (isn't a ready source of energy)

→ Proteins are taken to form living structure (isn't a ready source of energy) (to add new cells for muscles).etc. -es.

→ Carbohydrates are a ready source of energy



(priority is given to carbo. → fats → proteins)

→ a condition of prolonged starvation, a body requires regular intake of energy hence carbohydrate stored is used.

→ after carbo. we use fats.

→ after fats we use proteins.

Date:

### CARBOHYDRATES REQUIREMENTS FOR LIVING ORGANISMS:

- ready source of energy
- extra carbo. are stored in energy reserves in form of starch, glycogen  
in plants ↑ in animals ↑  
e.g/ pasta, rice, potatoes, sugar, vegetables

### PROTEIN REQUIREMENTS FOR LIVING ORGANISMS:

- energy carrying molecules
- are involved in formation of biomass. Eg: muscles, chemicals in body  
in animals ↑ enzyme, hormones, receptor ↑
- used as last priority in absence of carbohydrates and fats available for cellular respiration  
e.g/ meat, fish, eggs, peas, beans, lentils, soya

### FATS REQUIREMENT FOR LIVING ORGANISMS:

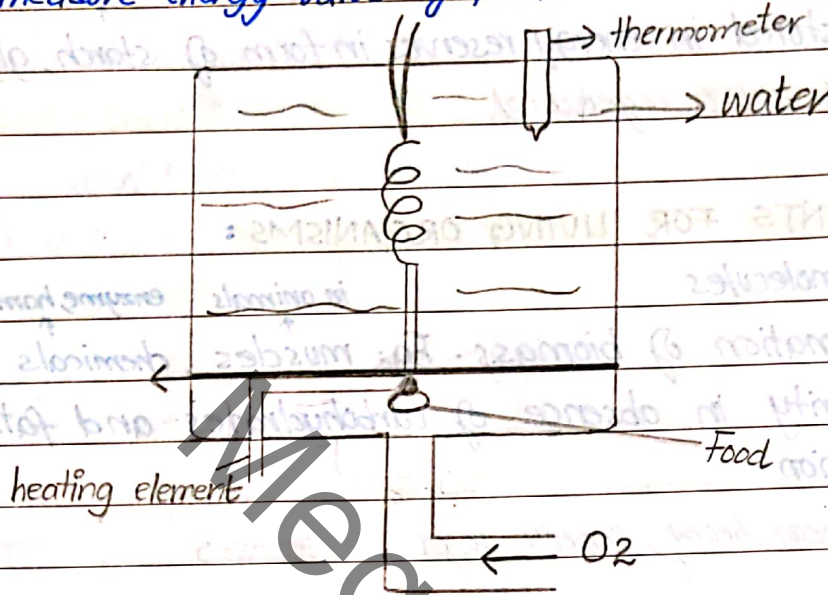
- are energy storage molecules
- extra fats are stored either in the form of glycogen or as adipose tissue underneath the skin
- " " " they provide us insulation (extra heat remains inside the body against extreme low temperatures outside)
- protective cushions around delicate organs. Eg: kidneys, ovaries
- plants store fats in form of reserved food in their seeds  
e.g/ cheese, butter, margarine, oil, nuts

### TEST FOR AMOUNT OF ENERGY:

- Take a specific food material on a spatula
- Heat it
- Heat energy released is calculated by placing a beaker and thermometer
- Rise in temperature is the amount
- more chances of error because gaps between the spatula and beaker will not give us exact amount of energy
- so a calorie or joules meter.

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- calorimeter is also called bomb-calorie meter
- used to measure energy value of foods



$$\text{Energy level of food} = \frac{\text{Rise in temperature (}^\circ\text{C)} \times \text{Volume of water (cm}^3\text{)}}{\text{Amount of food sample (g)}}$$

$$[\text{Joules (J)}] = A \text{ cal./grams} \text{ or } A \text{ } \frac{\text{joules}}{\text{gram}}$$

$$1 \text{ calorie} = 4.2 \text{ joules}$$

$$\text{Energy value of food in kJ} = \frac{A \times 4.2}{100}$$

→ Recommended Daily Allowance gives us a condition of balanced diet.  
on wrappers

### FACTORS AFFECTING BALANCED DIET

- ①. Age : infants need more ready source of energy. (carbo. + prot. + minerals)  
middle aged less ready source energy (physical activity only no cell division)  
old aged least because resting
- ②. Gender : males of same age require more energy as compared to a girl of same age because of a larger sized body
- ③. Occupation: (amount of physical energy) an office worker less energy  
a labour requires more

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- ④ Physiological Condition: (physical disorder) diabetic requires a diet with  
diabetic conditions → less carbohydrates  
high blood pressure → less fats  
anaemic condition → iron rich diet

$$\text{Balanced Diet} = \text{Basal Metabolic Rate/Index (BMR)} + \text{Physical} \\ + \text{Food Action}$$

BMR → amount of energy required to perform basic functions / metabolic activity  
Ex: circulatory system, respiratory system, excretory system

P.E → daily work, exercise evident to us

F.A → amount of energy required for digestion of food

1 gram of Fats ——— 39 KJ

1 gram of Protein ——— 20 KJ

1 gram of Carbohydrates ——— 17

- \* the higher the person's mass, the more energy they need.
- \* children require comparatively lesser amount

Date:

## Improving Foods (Modification of food)

### FORTIFICATION:

- Adding any 1 nutrient to increase the food value
- E.g.// fortified milk (NIDO), wheat (+ fluoride), water (+ fluoride), mineral water (+ salts in distillation) strengthens outer layer of teeth  
↓  
stay down during distillation

### ADDITIVES:

- Adding different chemicals to improve different conditions of food that not lead to enough nutrient level
- E.g.// • colouring agents (improving colour) رنگ دہکھانے والے  
↓  
tartazine, B-carotene
- flavouring agents (give flavour), vanilla essence, monosodium glutamate (MSG)
- preservatives (improving decomposition), antioxidants, sulphur dioxide
- thickeners (improving texture), cellulose (cornflour), Xanthan

\* not always beneficial. They cause side by side harms. ← excessive use

- tartazine —
- sulphur dioxide — sensitivity in asthmatic condition
- monosodium glutamate — nausea, vomiting
- freshwater + fertilizer — unfit for drinking  
↓  
we still have an indirect use
- pesticides are absorbed by plants — not readily degraded,  
↓  
toxic chemicals converted less harmful and excreted

(decreases population of animals in the end of food chain) thinning egg shells of birds which break before hatching, mental and nervous condition is affected

Date:

## BALANCED DIET

### MALNUTRITION

↓  
Bad

- i more food consumption than required
- ii very less food consumption than required
- iii unbalanced food consumption (1 nutrient more than required)

- i extra food than immediate need is stored in the body at different level.  
E.g.//
  - extra carbo. → glycogen → liver cells
  - extra proteins → amino acids → (degraded) deamination  
↓  
more urea → urine → out of body
  - forming muscles and string
  - extra fats → fatty acids & lipids → Adipose tissue under skin

→ overweight condition ⇒ OBESITY

- ii leads to deficiencies of certain specific nutrients.

E.g.//  
Vitamin C → scurvy condition  
Vitamin D → rickets  
Iron → anaemia

### ★ causes Kwashiorkor

↑ Protein deficiency — less use of mother feed (infants)

they are consuming more carbohydrate diet.

→ lack of muscles as a result

→ very thin arms and thin legs and lower body enlarges

### ★ Marasmus

overall diet deficiency (all nutrients) — less amount of food

→ only staying alive

DALMATIAN

**Date:**

→ very lean body

→ one of the cause is overpopulation / uneven distribution of food /  
advancement in medical (decreased death rate) / ~~pollution~~ (dry conditions,  
less rainfall, low water table) / flooding (fresh water reserves should be  
conserved) / global warming → causing abrupt melting

MegaLecture