TOPIC 7 TEST MARK SCHEME

1. (a) (i) fractional distillation or fractionation
(ii) $\mathrm{C}_{9} \mathrm{H}_{20}$ only $\rightarrow$
(iii) $\mathrm{C}_{11} \mathrm{H}_{24}+17 \mathrm{O}_{\overrightarrow{2}} \quad 11 \mathrm{CO}_{2}+12 \mathrm{H}_{2} \mathrm{O}$
(iv) $\mathrm{C}_{11} \mathrm{H}_{24}+6 \mathrm{O}_{2} \quad 11 \mathrm{C}+12 \mathrm{H}_{2} \mathrm{O}$
(b) (i) $\mathrm{C}_{10} \mathrm{H}_{22} \quad \mathrm{C}_{3} \mathrm{H}_{6}+\mathrm{C}_{7} \mathrm{H}_{16}$
(ii) correctly drawn structure of methylpropene (insist on clearly drawn $\mathrm{C}-\mathrm{C}$ and $\mathrm{C}=\mathrm{C}$ bords)
(c) Any two from
o chemically similar or chemically the same or react in the same way
o same functional group
o same general formula

- differ by $\mathrm{CH}_{2}$
(penarise same molecular formula or same empirieal formula)

2. (a) Structures of $\mathbf{Q}$ and $\mathbf{R}$ :


Q and $R$ in any order
3.


Allow e in the names
(ii) Structural (1) OR chain and position(al)
4.

Ml credit a correctsiructure for either geometrical $\mathrm{E}-\mathrm{Z}$ isomer and its designation as ejther cis or trans.
OR credit twio correct geometrical E-Z isomer structures (ignore the names)
OR credit two correct names for cis pent-2-ene and trans peni-a ene (ignore the structures)

M2 credit a second mark if all four parts of the required structures and names are correct.
(credit "linear" structures)
(insist on the alkyl groups being attached
clearly by C-C bonds)
5. (a) - (Same) General formula/allow a named homologous series with its general formula

- Chemically similar/same (chemical) reactions
- Same functional group
- Trend in physical properties/eg inc bp as $M_{r}$ increases
- (Molecules) increase by $\mathrm{CH}_{2} / \mathrm{M}_{\mathrm{r}}=14$

Any two points
(b) Fractional distillation/fractionation/chromatography Allow GLC
(c) (Molecules/compounds/substances) with the same melebular formula/same number and type of atoms

Allow alkanes with same molecular CSrmula
Allow same chemical formula in $1=0$ but can allow M2
but different structural formula/diffeert displayed formula/different arrangement of atoms/different tructures

Not different positions in space

2,4-dimethylhexane
M2 deperment on M1
$\mathrm{C}_{4} \mathrm{H}_{5}$
Ignore the absence of dash and/or commas
(d) less surface contact/less surface area/less polarisable molecule
so fewer/weaker/less Van der Waals'/vdw forces
Allow more spherical or fewer points of contact
Not smaller molecule/not more compact molecule/not shorter chain
Allow converse arguments
Must be comparative answer ie not just few VDW forces

QoL
Assume 'it' refers to the branched isomer
6. (a) 2-bromo-2,3-dimethylbutane

Ignore punctuation.
$\mathrm{C}_{n} \mathrm{H}_{2 n+1} \mathrm{Br}$ or $\mathrm{C}_{n} \mathrm{H}_{2 n+1} \mathrm{X}$ or $\mathrm{C}_{x} \mathrm{H}_{2 \times+1} \mathrm{Br}$
Any order.
1

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Stronger / more vdw (forces) between molecules (of 1bromohexane)

QoL
Allow converse arguments for Z
Not just more IMF.
Ignore size of molecule.
(b)

$\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{Cl}$
Any order
7. (a) (i) (Compounds with the) same molecular formula

Allow same number and type of atom for M1
Ignore same general formula.

But different structural formula / different displayed formula / different structures / different skeletal formula

M2 dependent on M1
Not different positions of atoms / bonds in space.

1

1
(ii) But-2-ene

Allow but-2-ene.
Allow but 2 ene.
Ignore punctuation.

(iii) (2)-methylprop-(1)-ene

Do not allow 2-methyleprop-1-erí.
(iv)


Do not allow skeletal formulae.
Penalise missing H and missing C
(b) (i) $\mathrm{C}_{4} \mathrm{H}_{8}+2 \mathrm{O}_{2} \quad 4 \mathrm{C}+4 \mathrm{H}_{2} \mathrm{O}$

Accept multiples.
(ii) Exacerbates asthma / breathing problems / damages lungs
/ smog / smoke / global dimming Ignore toxic / pollutant / soot / carcinogen. Do not allow greenhouse effect / global warming / acid rain / ozone.
(c) (i) $\mathrm{C}_{16} \mathrm{H}_{34}$

Allow $\mathrm{H}_{34} \mathrm{C}_{16}$
C and H must be upper case.
(ii) Jet fuel / diesel / (motor) fuel / lubricant / petrochemicals / kerosene / paraffin / central heating fuel / fuel oil

Ignore oil alone.
Not petrołł bitumen / wax / LPG / camping fuel.
(d) (i) $\mathrm{C}_{8} \mathrm{H}_{18}+25 \mathrm{NO} \quad 8 \mathrm{CO}_{2}+12.5 \mathrm{~N}_{2}+9 \mathrm{H}_{2} \mathrm{O}$

Accept multiples.
(ii) $\mathrm{Ir} /$ iridium

## OR

Pt / platinum

## OR

Pd / palladium

## OR

Rh / rhodium

8. (i) M1: compounds with the same structural formula

Mi2: but the bonds/groups/atoms have different spatial arrangements or orientation or configuration/are arranged differently in space/3D
(ignore reference to the same molecular formula for M1)
(ii) M1: correct structural representation for cis-but-2-ene and its name or its identification as the cis isomer

M2: correct structural representation for trans-but-2-ene and its name or its identification as the trans isomer

> (accept representations which are $90^{\circ}$ to linear)
> (award one mark for two correct structures but either wrong/no names)
> (maximum 1 mark for an incorrect alkene)
(iii) geometric(al) or cis-trans
9. B
10. B
[1]
11. C ..... [1]
12. C ..... [1]
13. C

