MEGA LECTURE

## TOPIC 7 TEST MARK SCHEME

1.	(a)	(i) fractional distillation or fractionation	1	
		(ii) $C_{9}H_{20} \text{ only } \rightarrow$	1	
		(iii) $C_{11}H_{24} + 17O_{2} + 11CO_{2} + 12H_{2}O$	1	
		(iv) $C_{11}H_{24} \rightarrow 6O_2$ 11C + 12H <sub>2</sub> O	1	
	(b)	(i) $C_{10}H_{22}$ $C_{3}H_{6} + C_{7}H_{16}$	1	
		(ii) correctly drawn structure of methylpropene (insist on clearly drawn C-C and C=C bonds)	1	
	(c)	Any two from		
		o chemically similar or chemically the same or react in the same way		
		o same functional group		
		o same general formula		
		o differ by CH <sub>2</sub> (penalise same molecular formula or same		
		empirical formula)	2 [8]	
2.	(a)	Structures of Q and R: $CH_3$ $C = C$ $CH_3$ and $CH_3$ $C = C$ $H$ $H$ $C = C$ $CH(CH_3)_2$		
		NOT C <sub>3</sub> H <sub>7</sub>		
		(1) (1)		
	Q and R in any order			
		1	[2]	

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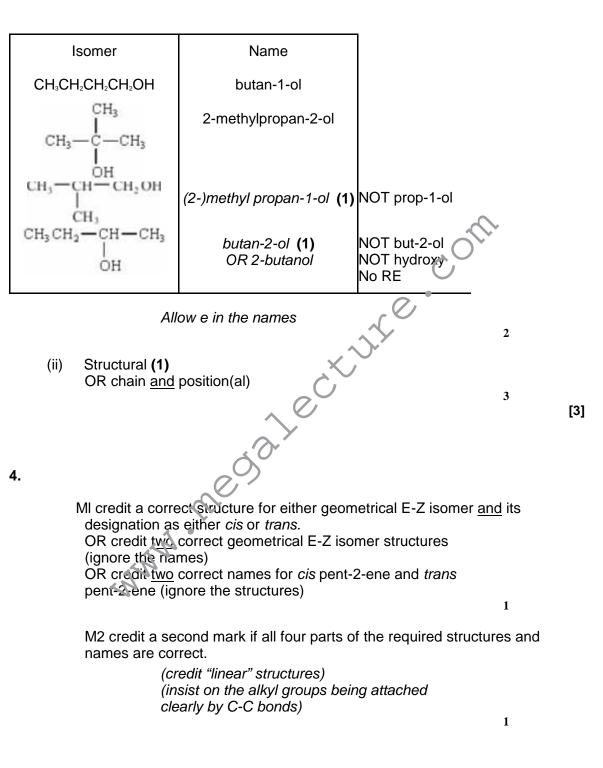


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3.



3

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5.	(a)	•	(Same) General formula/allow a named homologous series
			with its general formula

- Chemically similar/same (chemical) reactions
- Same functional group
- <u>Trend</u> in physical properties/eg inc bp as M<sub>r</sub> increases
- (Molecules) increase by CH<sub>2</sub>/M<sub>r</sub> = 14 Any two points
- (b) <u>Fractional</u> distillation/fractionation/chromatography Allow GLC
- (c) (Molecules/compounds/substances) with the same <u>molecular</u> formula/same number and type of atoms

Allow alkanes with same molecular formula Allow same chemical formula in M1 = 0 but can allow M2 2

1

1

1

1

1

but different structural formula/different displayed formula/different arrangement of atoms/different structures

Not different positions in space

2,4-dimethylhexane M2 dependent on M1

C₄H, Ignore the absence of dash and/or commas

- (d) less surface contact/less surface area/less polarisable molecule
  - so fewer/weaker/less <u>Van der Waals'/vdw</u> forces Allow more spherical or fewer points of contact Not smaller molecule/not more compact molecule/not shorter chain Allow converse arguments Must be comparative answer ie not just few

Must be comparative answer ie not just fev VDW forces





QoL Assume 'it' refers to the branched isomer

[9]

1

1

1

1

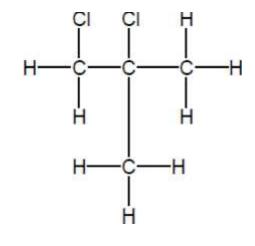
6. (a) 2-bromo-2,3-dimethylbutane Ignore punctuation.

> $C_nH_{2n+1}Br$  or  $C_nH_{2n+1}X$  or  $C_xH_{2x+1}Br$ Any order.

Stronger / more vdw (forces) between molecules (of 1bromohexane) QoL

Allow converse arguments for Z Not just more IMF. Ignore size of molecule.

(b)



 $C_2H_4CI$ 

Any order

[5]

1

1



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**7.** (a) (i) (Compounds with the) same molecular formula Allow same number and type of atom for M1 Ignore same general formula. 1 But different structural formula / different displayed formula / different structures / different skeletal formula M2 dependent on M1 Not different positions of atoms / bonds in space. 1 (ii) But-2-ene Joinylprop-(1)-ene Do not allow 2-methyleprop-1-ene. Allow but-2-ene. 1 (iii) (2)-methylprop-(1)-ene 1 e.c.X. (iv) н Do not allow skeletal formulae. Penalise missing H and missing C 1 (b) (i)  $C_4H_8 + 2O_2$ 4C + 4H<sub>2</sub>O Accept multiples. 1 (ii) Exacerbates asthma / breathing problems / damages lungs

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/ smog / smoke / global dimming Ignore toxic / pollutant / soot / carcinogen. Do not allow greenhouse effect / global warming / acid rain / ozone.



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	(c)	(i)	C <sub>16</sub> H <sub>34</sub> Allow H <sub>34</sub> C <sub>16</sub> C and H must be upper case.	1	
		(ii)	Jet fuel / diesel / (motor) fuel / lubricant / petrochemicals / kerosene / paraffin / central heating fuel / fuel oil Ignore oil alone. Not petrol-/ bitumen / wax / LPG / camping fuel.	1	
	(d)	(i)	$C_8H_{18} + 25NO = 8CO_2 + 12.5 N_2 + 9H_2O$ Accept multiples.	1	
		(ii)	Ir / iridium OR Pt / platinum OR Pd / palladium OR Rh / rhodium	1	
			Pt / platinum		
			OR OR		
			Pd / palladium		
			OR		
			Rh / rhodium		
				1	[11]
8.		(i)	M1: compounds with the same structural formula	1	
			2: but the bonds/groups/atoms have different spatial arrangements or orientation or configuration/are arranged differently in space/3D		
			(ignore reference to the same molecular formula for M1)		
			,	1	
		(ii)	<b>M1</b> : correct structural representation for cis-but-2-ene and its name or its identification as the cis isomer	1	
			<b>M2</b> : correct structural representation for trans-but-2-ene and its name or its identification as the trans isomer		
			9		



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(accept representations which are 90° to linear) (award one mark for two correct structures but either wrong/no names) (maximum 1 mark for an incorrect alkene)

1

		(iii)	geometric(al) or cis-trans	_	
•	-			1	[5]
9.	В				[1]
10.	В				
11.	С				[1]
					[1]
12.	С				[1]
13.	С				
					[1]



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