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r or example in the rest of the second of the source of th For carby the the there are the consider to pe coured use of the pate Bool of its calevant to this guestion m³. What are the values of x, y and z in a balanced equation? 72.0dm³ Ð $\label{eq:second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-second-seco$ କ୍ଷ A 100.1 z ().9009 **C 3 4 3 3 6 3 0 9 B** 12 9 m V161 6 **80 G**m³ 40 GM³ Ð Y DALSE BUR SIDE AN g) l as its active ingredient. The contains socium chlorate(II), NaCiO, nation of the second 2 **HERESEXTRAGEREN**ERTOXIDE, HZOZ. d 2// The diagram shows the mass spectrum of a sample of zinc. Use the data to calculate the relative $\begin{array}{c} \hline \label{eq:product} \hline \end{tabular} \\ \hline$ When 25.0 Gm^3 of bleggen is treated with as expess of a queous Hz2, 0.0350 mol of oxygen gas is a use of the system of 9401/01**4**01/04 A small hit at 16% best of a small vcantuk harabeseding baran kever ibate 3 rs, A, B, S, And D. Shoose the one you consider to → 2Na + 3N₂ question there are Feγr 1.5 1/be d $\cdot 2KNO_3 \rightarrow K_2O +$ $0 + N_2$ 10Na G/Vhates magha hgemente or the merogen i <u> ከ</u>ጠቀም 0.1 the number of nitros relevant to this quest op m 1 mol of 1 The anity of the broduced is used in the 10 The machine of an element between lithium and neon in the Periodic Table as follows?" 3 What is the maximum ampunt of TiO₂ obtainable from 19.0 tonnes of the ore illmenite, FeTiO₃? 49 1310²¹⁰ 3390 5320 7450 11 000 13 300 71 000 kJ mol⁻¹ **A** 10.0 tonnes **B** 12.7 tonnes **C** 14.0 tonnes **D** 17.7 tonnes y KJMOP^{⊢1} Mes What is the outer electronic configuration of the element? The twa sivelistic haps winkelignay is due used twaters of damage on cheartwink house is national that, 114 2 Kallerid detrighter to the following of the second state of the second s Which statement about ${}^{201}_{81}$ TO 3^{201}_{81} TO 3^{201}_{81} TO 3^{201}_{81} CO 2^{2}_{8} A sample of 10 cm³ of methane thin have the same the sa volume of gas remaining, the product was treated with an excess of aqueous sodium hydroxide and a mould have be the bound of the transmission of the transmiss temperature?, under such conditions that carbon disulphide was gaseous. C The number of neutrons in one atom of this isotope is 201. 70 cm³ (C) hat were it an isotope of worth is an isotope of worth in the second states of the second sec 50 cm³ D SO2 volume of gas 4384 tvolume of gas after C adding NaOH (aq)/cm³ ()© UCI ES 2004 after burning/om3 we have U 1 U Ø 10 reacts 30 9761/07/1200/09 10 Malae & Stoighio 1 56 10 ® Y 10 30 Ft 50 5 10 3 ted as snown. The structure have come cleating only in an a provided in their ground states? 38 www.megalecture.com Priod of the Periodic Table contains the same number of excess of 2 MEGA LECTURE For Live Online Classes 2_{PH₃}

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What as the electron arrangement of the nitrogen ion in gallium nitride? He Auto Alan Booklet Orale and the good and a company of the second seco 1 are as follows p What is the maximum amount of TiO₂ obtainable from 19.0 tonnes of the ore ilmenite, FeTiO₃? **A** 1310⁻¹ 3390 5320 7450 11 000 13 300 71 000 kJ mol⁻¹ **A** 10.0 tonnes **B** 12.7 tonnes **C** 14.0 tonnes **D** 17.7 tonnes What is the outer electronic configuration of the element? The ware inveliging skinks layray is due used masses of an as one of the structure 194 Adding 1, detright later varies in pay get a grading to the following a gra 2 1 103 Which statement about $\frac{201}{81}$ T (32)A sample of the main an excess of aqueous sodium hydroxide addatine wild he better better the transmission of the second with an excession additional sound in the second sound sou hat were is an isotope of at the measure of the set of C 50 cm³ D 70 cm³ 9⊚ uclisezer ampound S207 is hydrolysed by water to produce sulfuric acid and oxygen only. Which volume of gyvgen, measured at room temperature and pressure, is evolved when 0.352g of S_2O_7 is hydrolysed? 9701/01/**20**/09 96 cm³Moles & Stoichiometry WS 30 **C** 48 cm³ 24 cm³ В 2H2509701/01/1/1/1/06 2(1) 1 50 V: 0.003 x 24000 = 72 cm⁸ © UGEES 2006 + 2 H2 O \rightarrow 50 White Bair of Both atoms have one electron only in an s orbital in their ground states? 176.2 Ç P Li, Gr N 02 = MG1. 0.0544 Use of the Data Booklet is relevant to this question. Hard water contains calcum junspand average action at a single formation of a second calcumant in the second calcumant is the second calcumant in the second calcumant is the second calcumant in the second calcumant is the second calcumation of the second 10 How change decisions are present in the hydrogen carbon are diffion?, is produced when 0.783g of anhydrous barium nitrate reacts with an excess of aluminium? cm³ (B) $^{37}_{72.0 \text{ cm}^3}$ (C) $^{37}_{93.6 \text{ cm}^3}$ (D) $^{333}_{144 \text{ cm}^3}$ ($OAI + 3Ba(NO_3)_2 \longrightarrow 5AI_2O_3 + 3BaO + 3N_2$ **39**0 46.8 cm³ $\eta Ba(N_{3})_{2} = 0.783 = 0.003 \text{ mol.}$ 137 + 2(14 + 48)CHELES 2000.003 mol. \$701/01/QAN/08 $V \circ f N_2 = \frac{3}{1000} \times 24,000 = 72 \text{ cm}^3$

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