Nuclear Physics - 2018

1. 9702/11/M/J/18/No.38

Which elementary particle is a lepton?

- A proton
- B neutron
- c electron
- **D** quark
- 2. 9702/11/M/J/18/No.39

How many down quarks are in a nucleus of hydrogen-3, ³H?

- A 2
- B 3
- C

) 5 2 neutrons - udd + udd

3. 9702/11/M/J/18/No.40

What is the correct equation for β^+ decay?

- A neutron → proton + electron + electron antineutrino
- B neutron → proton + electron + electron neutrino
- C proton → neutron + positron + elegtron Antineutrino
- D proton → neutron + positron + electron neutrino
- proton neutra position neutra

4. 9702/12/M/J/18/No.38

In the α -particle scattering experiment, a beam of α -particles is aimed at a thin gold foil. Most of the α -particles go straight brough or are deflected by a small angle. A very small proportion are deflected through more than 90°, effectively rebounding towards the source of the α -particles.

Which conclusion about the structure of atoms cannot be drawn from this experiment alone?

- A Most of the atom is empty space.
- B Most of the mass of an atom is concentrated in the nucleus.
- The nucleus contains both protons and neutrons.
- D The nucleus is charged.

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5. 9702/12/M/J/18/No.39

Radon-211, $^{211}_{86}$ Rn, francium-210, $^{210}_{87}$ Fr, and radium-212, $^{212}_{88}$ Ra, are three nuclides.

How many neutrons does each nuclide have in its nucleus?

	radon-211	francium-210	radium-212	
A	86	87	88	
B	125	123	124	
С	211	210	212	
D	297	297	300	

$$R_n$$
, $n = 211 - 86 = 125$
 F_r , $n = 216 - 87 = 123$
 R_n , $n = 212 - 88 = 124.$

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A neutron is composed of one up (u) quark and two down (d) quarks. When the neutron decays to a proton, there is β -emission.

What is the change in the quark structure of the neutron due to the β-emission?

(The symbol for a neutrino is ν_{e} and for an antineutrino is $\overline{\nu_{e}}$.)

A d
$$\rightarrow$$
 u + β^- + ν_e

$$B$$
 d \rightarrow u + β^- + v_e

C
$$u \rightarrow d + \beta^+ + \nu_e$$

D u
$$\rightarrow$$
 d + β^{+} + ν_{e}

7. 9702/13/M/J/18/No.39

A nucleus of magnesium-23 undergoes β^{+} decay, as represented by the nuclear equation shown.

$$^{23}_{12} Mg \, \rightarrow \, X \, + \, \beta^{+} \, + \, \nu_{e}$$

What is nucleus X?

8. 9702/13/M/J/18/No.40

Which list contains only leptons?

(A) electron, neutrino, positron

B electron, neutrino, proton

electron, proton, neutron

neutrino, neutron, positron

Two sub-atomic particles types hadrone - protons, neutrons, pit, phi Leptons - electrons, neutrino, positron

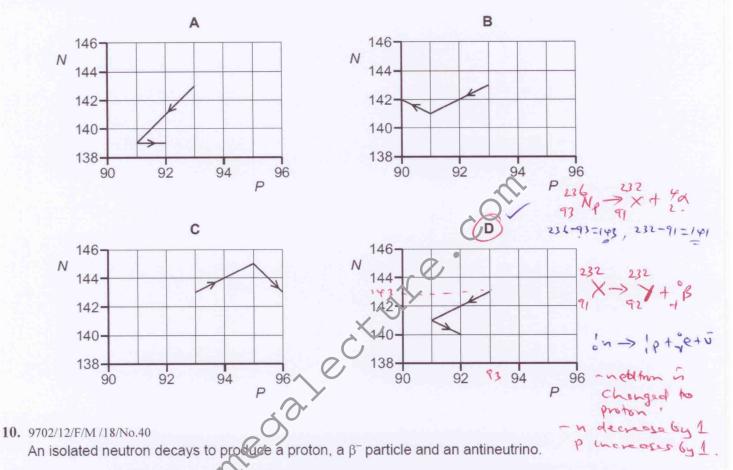
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9. 9702/12/F/M /18/No.39

A nucleus of neptunium-236 contains 93 protons and 143 neutrons. This nucleus decays with the emission of an α -particle. The nucleus formed then emits a β^- particle.

Which diagram shows the changes in the number P of protons and the number N of neutrons in these nuclei?



Which row gives the quark composition of the neutron and the proton and the type of force that gives rise to this reaction?

		n > p+B+5		
	quark composition		type of force	
	neutron	proton	type of force	udd > uud.
A	/ down, down, up	down, up, up	strong interaction	
B	down, down, up	down, up, up	weak interaction	-B deay is due to weak interaction
С	down, up, up	down, down, up	strong interaction	force.
D	down, up, up	down, down, up	weak interaction	

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